Magnetic Susceptibility of Insulating and Semiconducting Strontium Titanate

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The magnetic susceptibility of strontium titanate with and without charge carriers has been measured at 300, 78, and 4.2°K. Insulating SrTiO₃ shows a magnetic moment, which is partly diamagnetic and partly (Van Vleck) paramagnetic. The free-carrier contribution is mainly paramagnetic because of the high density of states. As the concentration of carriers is increased the temperature dependence disappears. At the highest concentration ($\sim 6 \times 10^{20}$ /cc), pure Pauli paramagnetism is observed. The density-of-states effective mass at the bottom of the conduction band deduced from these measurements is $5m_0$, in good agreement with values from transport experiments.

INTRODUCTION

ONIC salts without a magnetic moment in the ground state usually are diamagnetic. However, in many crystals contributions from higher energy states lead to an additional positive susceptibility (Van Vleck paramagnetism). Both parts are approximately temperature-independent. Occasionally the Van Vleck term is the larger of the two, as in the case of rutile (TiO_2) .¹

When conduction electrons are added to the crystaleither by doping the material or by thermal excitationa third term will appear resulting from the para- and diamagnetism of the free carriers.

This paper deals with SrTiO₃ in which all three terms can be observed and separated. The pure diamagnetic term for the unreduced material is simply the sum of the diamagnetic susceptibilities of the individual ions. The Van Vleck contribution depends on the energy band structure. The third term (due to free electrons) yields directly a value for the density-of-states effective mass at the bottom of the conduction band.

EXPERIMENTAL ASPECTS

The magnetic susceptibility was measured by the Faraday method using a high-sensitivity vertical-beam balance.²

A SrTiO₃ sample was cut from a single-crystal boule and ground into a small cylinder, 10 mm high and 3 mm in diameter, with a round bottom; the weight of this specimen was 0.3624 g. Experiments started with measurements on the fully oxidized material at 300, 78, and 4.2°K. Subsequently this sample was reduced three times and reoxidized once between the second and third reduction. These treatments are listed in Table I together with the carrier concentrations n. The latter are deduced from measurements of the Hall coefficient $(R_H = 1/nec)$ made on rectangular specimens that underwent the same reduction procedures.

The estimated uncertainty of the susceptibility measurements is $\pm 2\%$ or $\pm 5 \times 10^{-10}$ cm³ g⁻¹, whichever is larger.

Pure Strontium Titanate

Results on the fully oxidized sample of SrTiO3 are shown in the first and fourth rows of Table I. The lower negative value observed at 4.2°K is probably due to paramagnetic impurities like Fe, Cr, etc. As mentioned above, the susceptibility of an insulating crystal consists of a pure diamagnetic term and a term representing Van Vleck paramagnetism³:

$$\chi_{m} = -\frac{Le^{2}}{6mc^{2}} \sum_{i} \langle r_{i}^{2} \rangle_{av} + 2L \sum_{n,n'} \frac{|\langle n|\mu_{z}|n'\rangle|^{2}}{E_{n} - E_{n'}}.$$
 (1)

The "pure" diamagnetic part is supposed to be found by simply adding the diamagnetic susceptibilities of

Sample	Treatment	$n(4.2^{\circ}K)$ cm ⁻³	T _{deg} °K	χ or $\Delta \chi$ (10 ⁻⁷ cm ³ g ⁻¹)				
					300°K	78°K	4.2°K	$(m^*/m_0)_{4.2}$
Р	pure		•••	$\chi =$	-1.02	-1.00	-0.92	
R_1	28 h, 950°C hydrogen	6×1018	28	$\Delta \chi_1 =$	+0.037	+0.073	••••	
$egin{array}{c} R_1 \ R_2 \end{array}$	5 h, 1200°C hydrogen (carbon boat)	7.5×1019	148	$\Delta \chi_2 =$	+0.270	+0.523	+0.928	5.1
P'	reoxidized 18 h 700°C, air	•••	•••	<i>x</i> =	-1.016	-1.012	-0.919	•••
<i>R</i> ³	27 h, 1370°C hydrogen (carbon boat)	5.3×10 ²⁰	550	$\Delta \chi_3 =$	+1.704	+1.763	+1.719	4.9

TABLE I. Characteristics of a sample of SrTiO₃ before and after several hydrogen reductions.

¹ F. E. Senftle, T. Pankey, and F. A. Grant, Phys. Rev. **120**, 820 (1960). ² G. A. Candela and R. E. Mundy, Rev. Sci. Instr. **36**, 338 (1965).

⁸C. Kittel, Introduction to Solid State Physics (John Wiley & Sons, Inc., New York, 1956), p. 210.

the constituent ions. Using tabulated data⁴ for Sr²⁺⁻ $[(-17\pm2)\times10^{-6} \text{ cm}^3 \text{ mole}^{-1}], \text{Ti}^{4+}[(-5\pm2)\times10^{-6}]$ cm³ mole⁻¹], and O²-[$(-10\pm 2)\times 10^{-6}$ cm³ mole⁻¹], the first term is calculated to be $(-52\pm4)\times10^{-6}$ cm³ mole⁻¹. Subtracting this amount from the average experimental value of Table I $(-18.6 \times 10^{-6} \text{ cm}^3)$ mole-1), one finds a Van Vleck paramagnetism in $SrTiO_3$ of $(+33\pm4)\times10^{-6}$ cm³ mole⁻¹. When the same method is applied to the susceptibility of pure rutile (TiO_2) , $(+5.4 \times 10^{-6} \text{ cm}^3 \text{ mole}^{-1})$, the Van Vleck term appears to be $(+30\pm4)\times10^{-6}$ cm³ mole⁻¹. The close agreement between the Van Vleck paramagnetic contribution in these two substances is very satisfying considering the expected similarity in electronic band structure.⁵ It confirms the idea that the valence and conduction bands are pretty much determined by the $(TiO_6)^{2-}$ -octahedron.

Reduced Strontium Titanate

The magnetic susceptibility of the charge carriers in SrTiO₃ is obtained by determining the susceptibility of reduced samples of SrTiO₃ and subtracting the "pure" diamagnetic and Van Vleck contributions. The resulting values for 3 different concentrations of electrons (each at 300, 78, and 4.2°K) are presented in Table I.

Conduction electrons contribute both a diamagnetic moment (translational motion) and a paramagnetic moment (electron spin). In both cases the susceptibility depends on the prevailing statistics.

The total susceptibility can be written as follows⁶:

$$\chi(T) = \frac{n\mu^2}{kT} \frac{F_{1/2}'(\zeta)}{F_{1/2}(\zeta)} \left(1 - \frac{m^2}{3m^{*2}}\right), \qquad (2)$$

where n = number of charge carriers, $\mu = (g/2)\mu_B$, $\mu_B = Bohr$ magneton, $F_{1/2}$ and $F'_{1/2} = integral$ functions of $\zeta = E_F/kT(E_F = \text{Fermi energy})$, and $m^* = \text{density-of-}$ states effective mass at the bottom of the conduction band. One notices that the spin moment μ depends on the g factor. For conduction electrons g is usually (but not always) close to the free-electron value of 2.0023. Preliminary results from electron spin resonance experiments indicate that $g = 2.0 \pm 5\%$.⁷

For complete degeneracy the electronic susceptibility becomes independent of temperature (Pauli paramagnetism)⁶:

$$\chi(0) = (4m^*\mu^2/h^2)(3\pi^2n)^{1/3}(1-m^2/3m^{*2}).$$
(3)

⁴W. Klemm, Z. Anorg. Allgem. Chem. 244, 377 (1940); 246, 347 (1941); and P. W. Selwood, *Magnetochemistry* (John Wiley & Sons, Inc., New York, 1956), p. 78. ⁶A. H. Kahn and A. J. Leyendecker, Phys. Rev. 135, A1321

⁷ T. Chang (private communication).

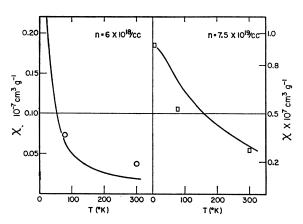
FIG. 1. Magnetic susceptibility of electrons in SrTiO₃ after the first and second reductions. The solid lines represent calculated values using Eq. (2).

The data of Table I show that Eq. (3) applies at all three temperatures only after the third reduction (R_3) . In this case the degeneracy temperature is considerably higher than 300°K. When the sample is less reduced $(R_1 \text{ and } R_2)$, Eq. (2) has to be used. However, all three samples are highly degenerate at 4.2°K and consequently Eq. (3) is applicable at this temperature. It should be noted that all parameters except m^* are known and hence it is possible to calculate the densityof-states effective mass. The values are listed in the last column of Table I. Considering the accuracy of the susceptibility measurement and the possible errors in g and n, these results are good to $\pm 8\%$. The value of m^* found in this experiment is in excellent agreement with the one derived from thermoelectric and Hall data.8 The temperature as well as the concentration dependence of m^* appears to be negligible.

It is now possible to check the susceptibilities measured at the two higher temperatures. Using our result $m^*=5m_0$, the carrier concentrations from our Hall data, and the tabulated values for $F_{1/2}(\zeta)$ and $F'_{1/2}(\zeta)$,⁹ the susceptibilities have been calculated between 4.2 and 300°K; these values together with the experimental data are shown in Fig. 1. It is seen that the temperature dependence is reasonably well represented by Eq. (2).

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^{(1964).}

⁶ A. H. Wilson, *The Theory of Metals* (Cambridge University Press, Cambridge, England, 1953), 2nd ed., pp. 155, 168.

⁸ H. P. R. Frederikse, W. R. Thurber, and W. R. Hosler, Phys. Rev. 134, A442 (1964).

⁹ O. Madelung, in *Handbuch der Physik*, edited by S. Flügge (Springer-Verlag, Berlin, 1957), Vol. 20, pp. 58, 59; $F_n'(\zeta) = nF_{n-1}(\zeta)$ for n > 0 (see Ref. 6, p. 332).